

## Revision lecture Phys1121-1131. UNSW

### Necessary tools:

- Vectors,  
components (force diagrams for Newton 2),  
addition and subtraction (relative velocities etc)

$$\mathbf{v} = \mathbf{v}_{\text{frame}} + \mathbf{v}'$$

dot product (for work)

$$\mathbf{a} \cdot \mathbf{b} = ab \cos \theta$$

$$\mathbf{a} \cdot \mathbf{b} = a_x b_x + a_y b_y$$

- Simultaneous equations  
solve n equations in n unknowns
- Simple differentiation and integration

### How big?

kg, m, s,  $\text{m.s}^{-1}$ ,  $\text{m.s}^{-2}$

N    kN    mN

J    kJ    mJ

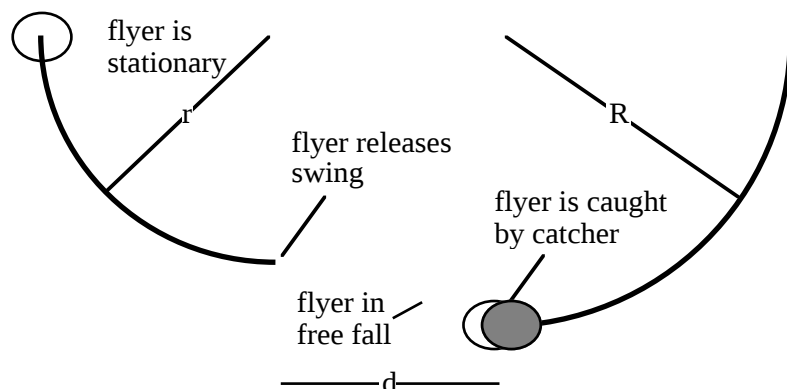
$\text{kg.m.s}^{-1}$      $10^3 \text{ kg.m.s}^{-1}$

K    kJ    ( $\text{kJ.kg}^{-1}.\text{K}^{-1}$ )

kW

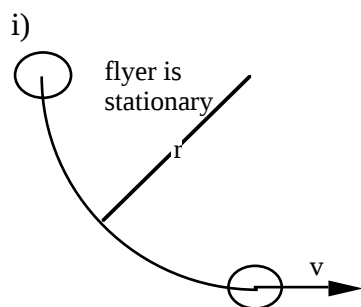
**Question.** A trapeze act in a circus is the one in which people swing and jump, high in the air. In a simple model of a simple trick, we treat the people as particles (ie neglect the internal motion and rotation), and neglect air resistance.

One person, called the flyer, with mass  $m$ , starts from rest on a swing that swings with radius  $r$ . Her initial motion is vertically down, as shown. When she is travelling horizontally, she releases the swing and travels in free fall for a horizontal distance  $d$ , at which point she is caught by the catcher, whose mass is  $M$ , and who is stationary at the instant of the catch. During the catch, their displacement is  $\ll R$ , where  $R$  is the radius of the arc in which they swing together after the catch. Determine how high they rise after the catch.



- i) Circular motion,  
what do we need? what principle?
- ii) Projectile motion  
what do we need? what principle?
- iii) Collision (elastic or not?)  
Is momentum conserved?  
So..... what do we need?
- iv) Circular motion  
what do we need? what principle?

- i) final  $v$  M.E. conserved
- ii) final  $v_x$   
a) (M.E. conserved.) b)  $v_x$  constant
- iii) Collision is **inelastic**. M.E. **not** conserved  
Momentum is *not* conserved in  $y$  direction  
Negligible external force in  $x$  direction so  
 $p_x$  is conserved.
- iv) M.E. conserved



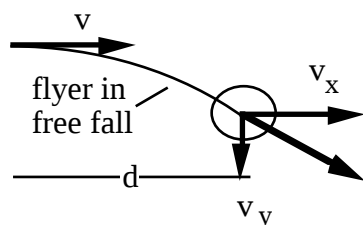
First swing: Non conservative forces do no work, so mechanical energy is conserved.

$$U_i + K_i = U_f + K_f$$

$$mgr + 0 = 0 + \frac{1}{2}mv^2$$

$$v = \sqrt{2gr}$$

ii)

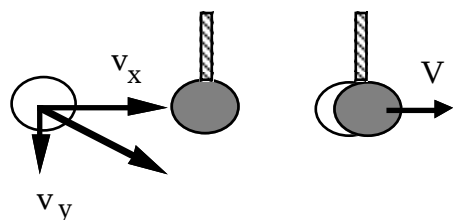


Projectile motion

$$v_x = v$$

(is all we need)

iii) Collision



Momentum **not** conserved in vertical direction: large external force in rope.

No external *horizontal* forces, so

$$p_{xi} = p_{xf}$$

$$mv_x = (M+m)V$$

$$V = \frac{mv}{M+m}$$

iv) No non conservative forces:

$$U_i + K_i = U_f + K_f$$

$$0 + \frac{1}{2}(M+m)V^2 = (M+m)gh + 0 \quad (\text{new zero for } U)$$

$$h = \frac{V^2}{2g} = \frac{1}{2g} \left( \frac{mv}{M+m} \right)^2 = r \left( \frac{m}{M+m} \right)^2$$

**Consolidate** with similar probs

Two or more steps (collision plus conservative steps)

- Ballistic pendulum (very similar)
- Collision and spring
  - Bullet proof vest
  - Hammering the chest



Collision

bullet hits vest

hammer hits brick

**p** conserved

Conservative step

compression of chest

compression of chest

K + U

$$\frac{1}{2}mv^2 + \frac{1}{2}kx^2 \text{ conserved}$$

## Revision of Heat

### Temperature (T) and heat Q:

T is equal in any 2 bodies at thermal equilibrium

T is intensive thermal energy  $k_B T/2$  in each molecular motion

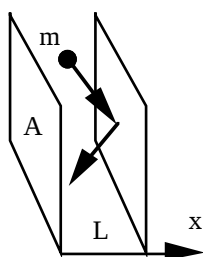
Q is extensive (total) thermal energy

( $\propto m$ :  $Q = cm\Delta T$ )

### Absolute temperature scale

$T \propto P$  in a low density gas

### Kinetic Theory



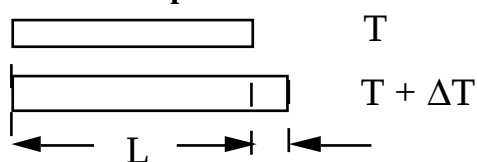
Newtonian calc  $\rightarrow$

$$PV = \frac{N}{3} m \overline{v^2}$$

$$P = \frac{1}{3} \rho \overline{v^2}$$

$$\therefore \quad \overline{\epsilon} = \frac{1}{2} m \overline{v^2} = \frac{3}{2} \frac{PV}{N} = \frac{3}{2} kT$$

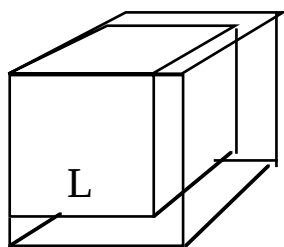
### Thermal Expansion



$$\therefore \quad \text{Define} \quad \frac{\Delta L}{L} = \alpha \Delta T \quad \frac{\Delta V}{V} = \beta \Delta T$$

$\alpha$  is coefficient of linear expansion

$\beta$  is coefficient of volume expansion



$$L + \Delta L$$

$$\Delta V = (L + \Delta L)^3 - L^3$$

$$= \dots$$

$$\approx V \cdot 3\alpha \Delta T$$

$$\therefore \beta \approx 3\alpha$$

Differential expansion problems (see notes)

**Example:** what is  $\beta$  for an ideal gas?

$$\beta \equiv \frac{1}{V} \left( \frac{\partial V}{\partial T} \right)_P \quad \text{and} \quad PV = nRT$$

$$V = \frac{nRT}{P}$$

$$\beta = \left( \frac{P}{nRT} \right) \frac{\partial}{\partial T} \left( \frac{nRT}{P} \right) = \frac{1}{T}$$

**Heat Capacity:** (for a body)  $C = \frac{\Delta Q}{\Delta T}$

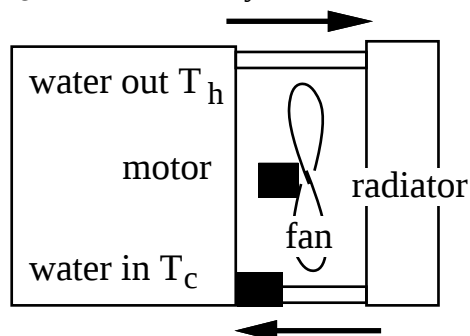
*extensive quantity*

**Specific Heat:** (of a substance)  $c = \frac{\Delta Q}{M \Delta T}$

*intensive quantity*

**Latent Heat:** heat required per unit mass for change of phase (at constant T).

**Question.** In steady state, T of motor is not changing.



Water in to radiator at 80°C and out at 50°C.

Motor generates 30 kW of heat. What is the flow rate through the radiator?

measure flow rate in kg/s:  $\frac{dm}{dt} = ?$  (i)

heat needed to raise T of water.

We are given  $\frac{dQ}{dt} = 30 \text{ kW}$  (ii)

What quantity do we need?

Specific heat  $c \equiv \frac{1}{m} \frac{Q}{\Delta T}$

Heat to warm  $dm$  of water  $dQ = dm \cdot c \Delta T$  (iii)

$$\begin{aligned} \frac{dm}{dt} &= \frac{1}{dt} \cdot \frac{dQ}{c \Delta T} = \frac{1}{c \Delta T} \frac{dQ}{dt} \\ &= \frac{1}{4.2 \text{ kJ} \cdot \text{kg}^{-1} \cdot \text{K}^{-1} \cdot 30 \text{ K}} 30 \text{ kW} \\ &= 0.24 \text{ kg/s} \quad (14 \text{ litres/min}) \end{aligned}$$

If motor is 33% efficient, how rapidly can it drive a 1 t car up a 1:10 hill (neglect air resistance)?

1/3 efficient. If 30 kW wasted, 15 kW produced. (45kW used)

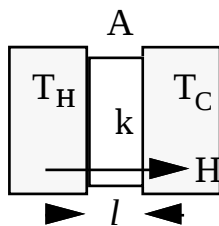
$$\frac{d}{dt} mgh = 15 \text{ kW}$$

$$\frac{dh}{dt} = \frac{15 \text{ kW}}{1000 \text{ kg } 9.8 \text{ ms}^{-2}} = 1.5 \text{ ms}^{-1}$$

On 1:10 hill,  $v = 15 \text{ m/s} = 54 \text{ kph}$  ( $>$  speed limit)

(at this speed air resistance  $<$  component of weight)

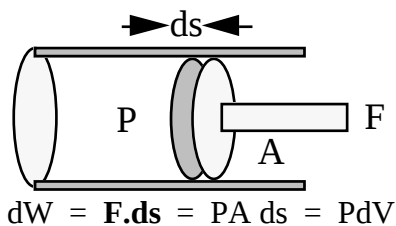
### Heat conduction



$$H \equiv kA \frac{T_H - T_C}{l}$$

In *steady state*, heat is lost at the same rate it is produced. (Several examples in notes)

**Work** done against pressure P



**1st Law**  $dU = dQ - dW$   
where U is a state function

**Isobaric** P constant

**Adiabatic Process:**  $\Delta Q = 0$  (fast or insulated)

**Isothermal process:** T constant

**Work out work:**

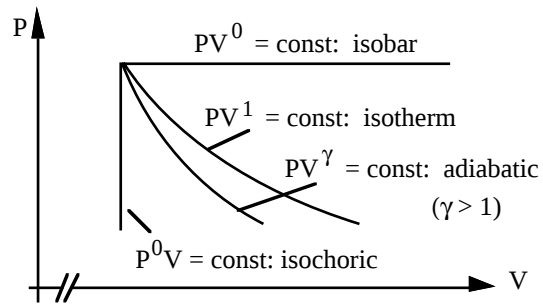
Isobaric:  $W = \int P \, dV = P\Delta V$

Isothermal:  $W = \int P \, dV = \int \frac{nRT}{V} \, dV = nRT \ln \frac{V_f}{V_i}$

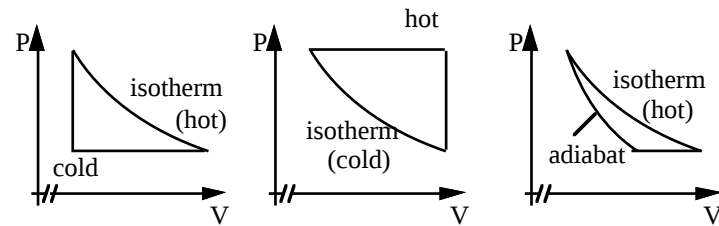
Adiabatic:  $PV^\gamma = \text{constant}$

$$\gamma \equiv \frac{c_P}{c_V} > 1 \quad (\sim 1.4 \text{ for air})$$

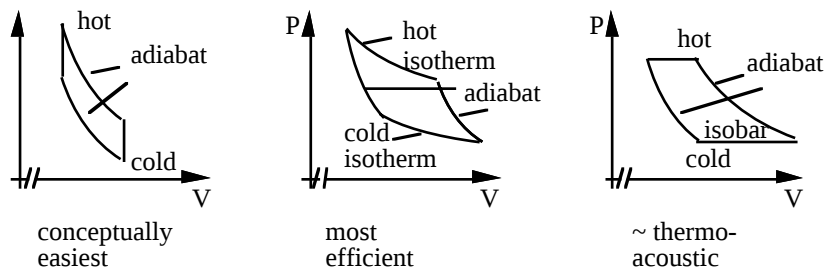
## P-V diagrams and thermal cycles



Simple cycles typical of those found in tut and exam questions



Idealised cycles sometimes approximated



What happens to  
isothermal  
expansion  
adiabatic  
expansion  
isobaric  
expansion

W,

U and

Q in

whole cycle